Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

Retraction of articles by H. Zhong *et al.*

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Received 20 November 2009; accepted 15 December 2009

A series of 41 papers by H. Zhong *et al.* are retracted.

As a result of problems with the data sets and incorrect atom assignments, 41 papers by H. Zhong *et al.* are retracted. Full details of all the articles are given in Table 1.

Table 1
Details of articles to be retracted, in order of publication.

Title	Reference	DOI	Refcode
Aquachlorobis(1,10-phenanthroline)cobalt(II) chloride thiourea solvate	Zhong, Zeng, Liu & Luo (2006*a*)	10.1107/S1600536806041122	KERQEE
cis-Dichlorobis(1,10-phenanthroline)cobalt(II)	Zhong, Zeng & Luo (2006)	10.1107/S1600536806047295	MEQFOE
Tris(quinolin-8-olato-κ^2N,O)cobalt(III) glyoxal hemisolvate monohydrate	Zhong, Zeng, Liu & Luo (2006*b*)	10.1107/S1600536806050240	MEQHEW
(8-Quinolinol-κ^2N,O)bis(8-quinolinolato-κ^2N,O)nickel(II) glyoxal hemisolvate monohydrate	Zhong, Zeng, Liu & Luo (2007)	10.1107/S1600536806053232	METVUD
Aquachlorobis(1,10-phenanthroline)cobalt(II) chloride thioacetamide solvate	Zhong, Zeng & Luo (2007)	10.1107/S1600536806053530	METQIM
(8-Quinolinol-κ^2N,O)-bis(8-quinolinolato-κ^2N,O)zinc(II) glyoxal hemisolvate monohydrate	Zhong, Zeng, Luo, Li & Xiao (2007)	10.1107/S1600536807001171	DEXTEG
(Dimethylglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')nickel(II) dinitrate dihydrate	Zhong, Zeng, Yang, Luo & Li (2007*a*)	10.1107/S1600536807004102	YEYGOZ
(Dimethylglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')zinc(II) dinitrate dihydrate	Zhong, Zeng, Yang, Luo & Li (2007*b*)	10.1107/S1600536807004096	YEYGUF
Chloridobis(1,10-phenanthroline-$\kappa N,N'$)copper(I) hexahydrate	Zhong, Zeng, Yang, Luo & Xiao (2007)	10.1107/S160053680700791X	HEGKOU01
Tetrakis(pyridine-κN)bis(thiocyanato-κN)cobalt(II)	Zhong, Zeng, Yang & Luo (2007*a*)	10.1107/S1600536807017461	ITCPCO01
Tetrakis(pyridine-κN)bis(thiocyanato-κN)copper(II)	Zhong, Zeng, Yang & Luo (2007*b*)	10.1107/S160053680701879X	AVUJEG02
Tetrakis(nitrato-κ^2O,O')bis(4-phenylpyridine-κN)cerium(IV)	Zhong, Zeng, Yang & Luo (2007*c*)	10.1107/S1600536807018831	CICDOI
Bis(4,4′-bipyridine-κ^2N,N')tetrakis(nitrato-κ^2O,O')cerium(IV)	Zhong, Zeng, Yang & Luo (2007*d*)	10.1107/S1600536807021502	YIDNEF
(1,10-Phenanthroline)tris(phenoxyacetato)lanthanum(III)	Zhong, Zeng, Yang, Luo & Xu (2007)	10.1107/S1600536807027171	EDUROL
(1,10-Phenanthroline)tris(phenoxyacetato)cerium(III)	Zhong, Yang, Luo & Xu (2007*a*)	10.1107/S1600536807028061	EDUTUT
(1,10-Phenanthroline)tri(3-phenylpropanoato)lanthanum(III)	Zhong, Yang, Luo & Xu (2007*b*)	10.1107/S1600536807028693	RIGQEE
(1,10-Phenanthroline-κ^2N,N')tris(phenoxyacetato)-$\kappa O;\kappa O;\kappa O,O'$-neodymium(III)	Zhong, Yang, Luo & Xu (2007*c*)	10.1107/S1600536807030371	UDUMEM
Bis(2,2′-bipyridyl-κ^2N,N')bis(thiocyanato-κN)nickel(II)	Zhong, Yang, Luo & Xu (2007*d*)	10.1107/S1600536807031613	YEJGOJ01
Bis(2,2′-bipyridyl-κ^2N,N')bis(isothiocyanato-κN)copper(II)	Zhong, Yang, Luo & Xu (2007*e*)	10.1107/S1600536807033181	UFAPOH
Bis(2,2′-bipyridyl-κ^2N,N')bis(thiocyanato-κN)zinc(II)	Zhong, Yang, Luo & Xu (2007*f*)	10.1107/S1600536807035337	TIGFAR
(1,10-Phenanthroline-κ^2N,N')tris(3-phenylpropanoato-κO)neodymium(III)	Zhong, Yang, Luo & Xu (2007*g*)	10.1107/S1600536807035350	TIGFEV
2-Fluoro-3,5-dinitrobenzamide monohydrate	Zhong, Yang, Xie & Luo (2007*j*)	10.1107/S1600536807038676	VIKGAY
2-Fluoro-3,5-dinitrobenzoic acid-ammonia (1/1)	Zhong, Yang, Xie & Luo (2007*k*)	10.1107/S1600536807039724	KILKIA
1-Hydroxy-4,6-dinitropyridine-2-carboxamide monohydrate	Zhong, Yang, Xie & Luo (2007*l*)	10.1107/S1600536807040779	AFETAH
N-(2-Hydroxyphenyl)carbamic acid-ammonia (1/1)	Zhong, Yang, Xie & Luo (2007*m*)	10.1107/S160053680704086X	AFINAF
catena-Poly[[bis(μ-anilinoacetato-$\kappa^2O:O'$)bis(μ-anilinoacetato-$\kappa^3O,O':O$)bis[(1,10-phenanthroline-κ^2N,N')samarium(III)]-μ-anilinoacetato-$\kappa^2O:O'$]	Zhong, Yang, Xie & Luo (2007*a*)	10.1107/S1600536807043528	PILDAQ
2-Hydroxy-5-nitrobenzene-1,3-dicarboxylic acid monohydrate	Zhong, Yang, Xie & Luo (2007*n*)	10.1107/S1600536807045199	XILWIZ
catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dineodymium(III)]-di-μ-anilinoacetato]	Zhong, Yang, Xie & Luo (2007*b*)	10.1107/S1600536807048489	WIMWEV
Hexaaquacopper(II) bis(4-methylbenzenesulfonate)	Zhong, Yang, Xie & Luo (2007*c*)	10.1107/S1600536807049525	TOLSCV01

Table 1 (continued)

Title	Reference	DOI	Refcode
catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dilanthanum(III)]-di-μ-anilinoacetato]	Zhong, Yang, Xie & Luo (2007*d*)	10.1107/S1600536807051240	GIMZEI
Hexaaquachromium(II) bis(4-methylbenzenesulfonate)	Zhong, Yang, Xie & Luo (2007*e*)	10.1107/S1600536807051227	GIMZIM
Hexaaquamanganese(II) bis(4-methylbenzenesulfonate)	Zhong, Yang, Xie & Luo (2007*f*)	10.1107/S1600536807052051	QUKQES01
catena-Poly[[(acetato-κO)(1,10-phenanthroline-κ^2N,N')cobalt(II)]-μ-acetato-$\kappa^2O:O'$]	Zhong, Yang, Xie & Luo (2007*g*)	10.1107/S1600536807053494	NIQLAB
Hexaaquanickel(II) bis(4-aminobenzenesulfonate)	Zhong, Zhong, Xie & Luo (2007*a*)	10.1107/S1600536807054372	HIPZOW
catena-Poly[[(acetato-κO)(1,10-phenanthroline-κ^2N,N')copper(II)]-μ-acetato-$\kappa^2O:O'$]	Zhong, Yang, Xie & Luo (2007*h*)	10.1107/S160053680705622X	XIRGOV
Hexaaquazinc(II) bis(4-aminobenzenesulfonate)	Zhong, Zhong, Xie & Luo (2007*b*)	10.1107/S1600536807056498	XIRJEO
catena-Poly[[(acetato-κO)(1,10-phenanthroline-κ^2N,N')nickel(II)]-μ-acetato-$\kappa^2O:O'$]	Zhong, Yang, Xie & Luo (2007*i*)	10.1107/S1600536807058540	HIQJOH
Hexaaquacobalt(II) bis(4-aminobenzenesulfonate)	Zhong, Xie & Luo (2007)	10.1107/S1600536807058527	HIQJUN
catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dieuropium(III)]-di-μ-anilinoacetato]	Zhong, Yang, Duan & Hong (2007)	10.1107/S1600536807060643	YIQMAN
(Dimethylglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')copper(II) dinitrate dihydrate	Zhong, Yang, Luo & Li (2007)	10.1107/S1600536807061193	YIQNUI
catena-Poly[[(1,10-phenanthroline-κ^2N,N')praseodymium(III)]-di-μ-phenoxyacetato-$\kappa^4O:O'$-[(1,10-phenanthroline-κ^2N,N')-praseodymium(III)]-di-μ-phenoxyacetato-$\kappa^4O:O'$-di-μ-phenoxyacetato-$\kappa^3O,O':O;\kappa^3O:O,O'$]	Zhong, Yang, Luo & Xu (2008)	10.1107/S1600536807068614	GISJIC

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Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

(8-Quinolinol-κ^2N,O)bis(8-quinolinolato-κ^2N,O)-zinc(II) glyoxal hemisolvate monohydrate

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Received 24 December 2006
Accepted 10 January 2007

The Zn^{II} atom in the title complex, $[Zn(C_9H_6NO)_2(C_9H_7NO)]\cdot0.5C_2H_2O_2\cdot H_2O$, exists in an octahedral coordination geometry defined by three O and three N atoms from three 8-hydroxyquinoline ligands. The glyoxal molecule lies on a centre of inversion. In the crystal structure, molecules are linked into a three-dimensional framework by O—H···O hydrogen bonds and π–π stacking interactions.

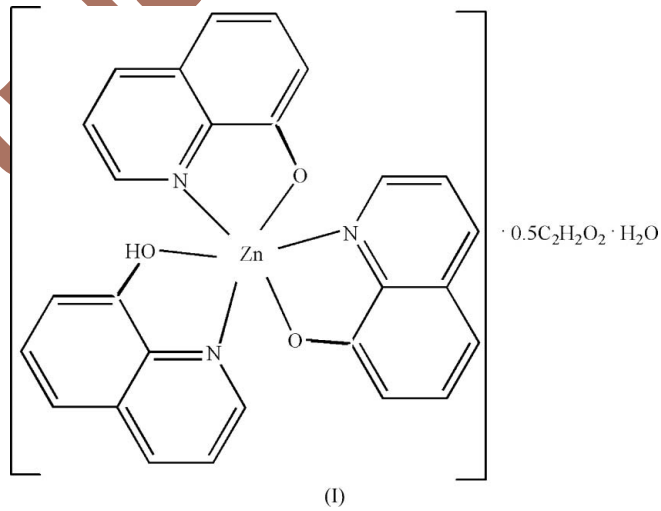
Key indicators

Single-crystal X-ray study
T = 273 K
Mean σ(C–C) = 0.008 Å
R factor = 0.045
wR factor = 0.175
Data-to-parameter ratio = 14.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Comment

In recent years, interest in the chemistry of metal–oxygen clusters has grown because of their applications in areas such as catalysis, materials chemistry and biochemistry (Pope, 1983; Pope & Müller, 2001). π–π Stacking between aromatic rings is related to the electron-transfer process in some biological systems (Deisenhofer & Michel, 1989; Wall *et al.*, 1999). Aromatic polycyclic compounds, such as quinoline, phenanthroline and benzimidazole, have commonly shown π–π stacking in metal complexes (Wu *et al.*, 2003; Pan & Xu, 2004; Liu *et al.*, 2004; Li *et al.*, 2005). These complexes can be used to develop new diagnostic and therapeutic agents in DNA binding and cleavage (Barton, 1986; Naing *et al.*, 1995). We report here the crystal structure of the title compound, (I).



(I)

In the molecule of (I) (Fig. 1), the three O atoms and three N atoms of three 8-hydroxyquinoline ligands are coordinated to the Zn^{II} atom in an octahedral arrangement (Table 1). The Zn—O bonds [average 1.901 (3) Å] are somewhat shorter than the Zn—N distances [average 1.927 (4) Å]. The planar 8-hydroxyquinoline ligands are nearly perpendicular to each other, with dihedral angles of 85.96 (5), 87.41 (4) and 89.96 (3)°. The glyoxal molecule lies on an inversion centre.

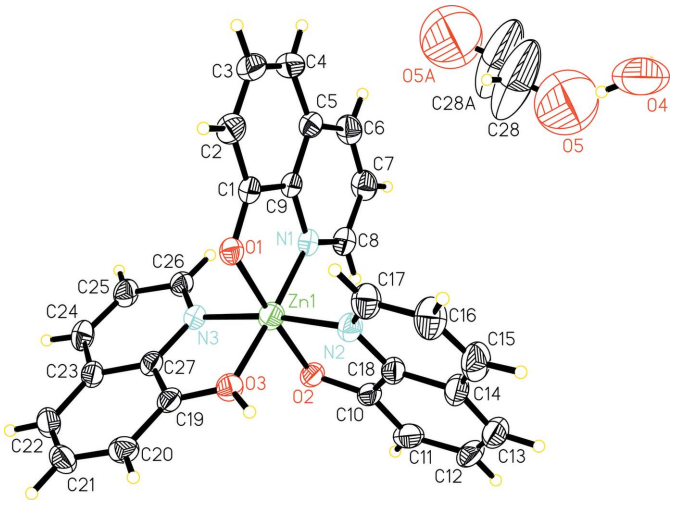


Figure 1
The molecular structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. Atoms labelled with the suffix A are generated by the symmetry operator $(1-x, 1-y, 1-z)$.

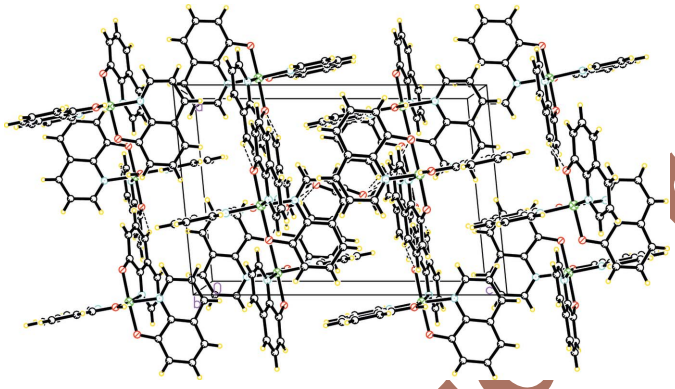


Figure 2
A packing diagram for (I). Hydrogen bonds are shown as dashed lines.

In the crystal structure, molecules are linked into a three-dimensional framework by O—H···O hydrogen bonds (Table 2). There are π–π stacking interactions involving the N3/C23–C27 (centroid *Cg*1) and N2/C14–C18 (centroid *Cg*2) pyridine rings of adjacent 8-hydroxyquinoline ligands, with $Cg1\cdots Cg1^{i}$ and $Cg2\cdots Cg2^{ii}$ distances of 3.609 (5) and 3.711 (5) Å, respectively [symmetry codes: (i) $1-x, 2-y, 2-z$; (ii) $1-x, 2-y, 1-z$]. These hydrogen bonds and π–π stacking interactions lead to a supramolecular network structure (Fig. 2).

Experimental

Zinc dinitrate hexahydrate (297 mg, 1 mmol), 8-hydroxyquinoline (290 mg, 2 mmol) and glyoxal (106 mg, 2 mmol) were dissolved in ethanol (25 ml). The mixture was heated for 10 h under reflux with stirring. It was then filtered to give a clear solution, into which diethyl ether vapour was allowed to condense in a closed vessel. After being allowed to stand for a few days at room temperature, colourless single crystals of (I) suitable for X-ray diffraction analysis precipitated.

Crystal data

$[Zn(C_9H_6NO)_2(C_9H_7NO)]\cdot 0.5C_2H_2O_2\cdot H_2O$
$M_r = 545.86$
Monoclinic, $P2_1/n$
$a = 11.2499$ (10) Å
$b = 12.8308$ (11) Å
$c = 16.7249$ (14) Å
$\beta = 95.588$ (1)°
$V = 2402.7$ (4) Å^3
$Z = 4$
$D_x = 1.509$ Mg m^{-3}
Mo $K\alpha$ radiation
$\mu = 1.07$ mm^{-1}
$T = 273$ (2) K
Prism, colourless
0.30 × 0.17 × 0.16 mm

Data collection

Bruker APEXII area-detector diffractometer
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
$T_{min} = 0.800$, $T_{max} = 0.848$
19259 measured reflections
4997 independent reflections
3036 reflections with $I > 2\sigma(I)$
$R_{int} = 0.039$
$\theta_{max} = 26.6°$

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.045$
$wR(F^2) = 0.175$
$S = 1.02$
4997 reflections
342 parameters
H atoms treated by a mixture of independent and constrained refinement
$w = 1/[\sigma^2(F_o^2) + (0.1P)^2 + 0.5743P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} = 0.001$
$\Delta\rho_{max} = 0.55$ e Å^{-3}
$\Delta\rho_{min} = -0.61$ e Å^{-3}

Table 1
Selected geometric parameters (Å, °).

Zn1—O1	1.901 (3)	Zn1—N1	1.922 (4)
Zn1—O2	1.910 (3)	Zn1—N2	1.925 (4)
Zn1—O3	1.891 (3)	Zn1—N3	1.934 (3)
O1—Zn1—O2	177.45 (12)	O2—Zn1—N3	90.39 (13)
O1—Zn1—O3	89.99 (13)	O3—Zn1—N1	175.35 (14)
O2—Zn1—O3	91.96 (13)	O3—Zn1—N2	87.91 (15)
O1—Zn1—N1	85.36 (13)	O3—Zn1—N3	85.58 (14)
O1—Zn1—N2	92.44 (14)	N1—Zn1—N2	92.23 (15)
O1—Zn1—N3	91.40 (14)	N1—Zn1—N3	94.57 (15)
O2—Zn1—N1	92.68 (14)	N2—Zn1—N3	172.44 (16)
O2—Zn1—N2	85.98 (14)		

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O4—H4*B*···O5	0.862 (10)	1.69 (3)	2.32 (2)	127 (3)

The H atoms of the water molecule were located in a difference synthesis and refined isotropically, with the O—H and H···H distances restrained to 0.84 (1) and 1.37 (2) Å, respectively. The remaining H atoms were positioned geometrically, with C—H = 0.93 Å, and constrained to ride on their parent atoms, with U_{iso}(H) = $1.2U_{eq}$(C).

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXTL*.

This work was supported by the Science and Technology Bureau of Jian, Jiangxi Province, China (grant No. 20052817).

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